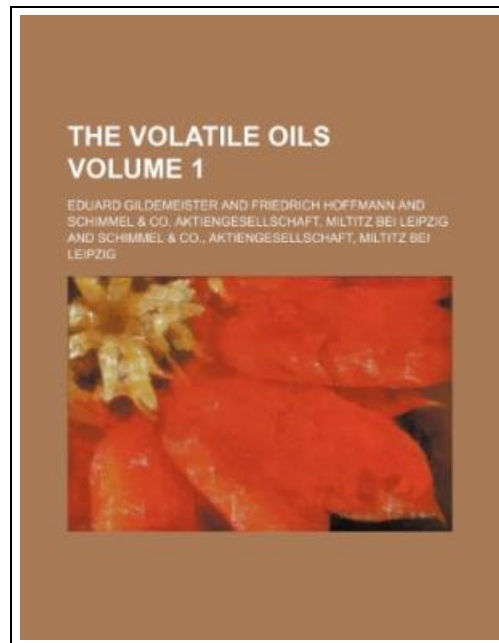


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Reviews

Thorough manual for publication fanatics. It is actually rally intriguing throug reading through period of time. Its been written in an remarkably simple way and is particularly only after i finished reading through this book in which actually transformed me, change the way i think.

(Morris Schultz)

THE VOLATILE OILS VOLUME 1



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1913 Excerpt: .and fennel, also in the oils of Satureja Thymbra, Thymus capitatus, in American pennyroyal oil, in kessoroot oil and in goldenrod oil.) Berl. Berichte 27 (1894), 448. a) Pharm. Rundschau (New York) 12 (1894), 160; Arch, der Pharm. 232 (1894), 646. 8) Chem. Ztg. 22 (1898), 827. Aside from its formation from equal parts of d-and /-limonene, it is obtained synthetically by polymerization of the unsaturated aliphatic hydrocarbon isoprene, C₅H₈; also, together with terpinene, upon dehydration of the aliphatic alcohols linalool and geraniol. Furthermore, it results upon isomerization of other hydrocarbons C₁₀H₁₆, e.g. pinene, limonene, phellandrene; also by inducing the proper changes in related oxygenated compounds, such as cineol, terpineol, and terpinhydrate. Dipentene, together with cineol and limonene, was also obtained by the action of nitrous acid on fenchylamine. A relatively pure dipentene can be obtained by the dry distillation of caoutchouc. After the removal of the isoprene, fraction 172 to 178 is carefully and repeatedly fractionated over sodium. The presence of any cineol, which has a like boiling point, can be ascertained by a corresponding increase in the specific gravity. A less pure preparation can be obtained from dipentene dihydrochloride by splitting off hydrogen chloride with aniline or sodium acetate in glacial acetic acid;1) also by the dehydration of crystalline terpineol by means of potassium acid sulphate. -) Physically, dipentene differs from limonene only by its optical inactivity. The boiling point of pure dipentene appears to be the) Wallach, Liebig s Annalen 239...



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