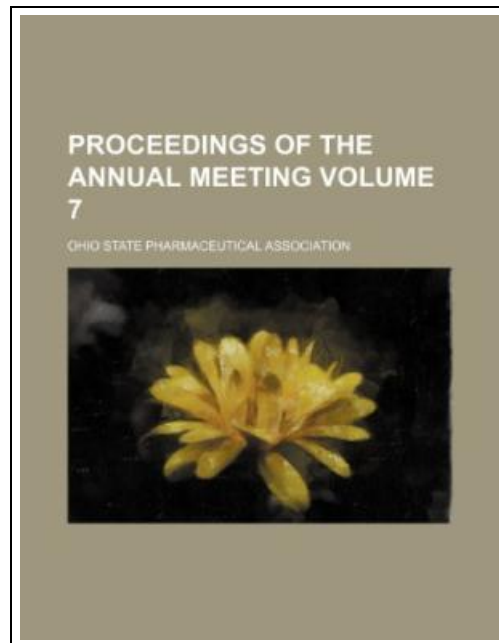


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Reviews

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Not Avail, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1885 Excerpt: . cannot be recommended in preparing emulsions, and with the field still open for further investigation, I remain, Respectfully, John Rauschkolb, Columbus, O. IODUM. MISS MARY H. SPENZER, Cleveland, O. Query No. 6.--An examination of Commercial Iodine is wanted. For the above Query, eleven samples of resublimed Iodine of commerce, labeled G. Mallinckrodt Co., Chas. Pfizer, Powers Weightman and Billings, Clapp Co., were subjected to the following tests: i. Appearance. 2. Odor. 3. Volatilization. 4. Solution in its Several Solvents--Water, Ether, Chloroform, etc. 5. Detection of Chlorine and Bromine. 6. Detection of Cyanide of Iodine. 7. Detection of Calcium Salts. 8. Estimation of Iodine. 9. Estimation of Water. 1. Appearance.--Several substances can be detected by this means: (a) Water, by the sticky coherence of the Iodine together or to the side of the bottle. (b) Non-volatile adulterations, such as coal, black antimony, etc., can be inferred. (c) By the aid of a magnifying glass, minute, colorless, needle-shaped crystals of Cyanide of Iodine can be detected. 2. Odor.--Iodine has not a very strong odor, therefore the very disagreeable odor of Cyanide of Iodine can be observed if present. 3. Volatilization.--Pure Iodine should completely volatilize, leaving no residue when heated in a test tube; if otherwise, it is adulterated; Cyanide of Iodine, if present, will be seen sublimed as colorless crystals above the Iodine. 4. Solution in its Several Solvents.--(a) The solution, in water, should be of a yellowish-brown color; while if Chloride of Iodine is present,...



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