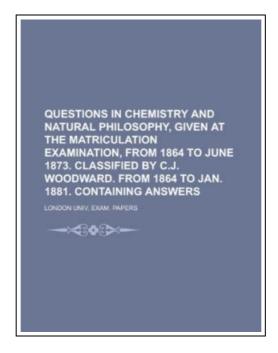
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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand ******. This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1881 edition. Excerpt: the process you would adopt for the determination of the specific heat of a metal. 1000 grammes of water are heated 0,114 C. by the heat given off when 100 grammes of iron are cooled from 20 to 10. Calculate from these data the specific heat of iron. 7. One kilogramme of ice at 0 C. and 3 kilogrammes water at 79 C. are mixed in a closed vessel the sides of which are supposed impervious to heat. What will be the temperature of the water after the melting of the ice? (Latent heat of water 79.) 8. You find by experiment that the heat of the flame of a single spirit-lamp is sufficient to boil 6 ounces of water at 100 C. in an open evaporating dish. Two flames, each having the same size and temperature as the one first mentioned, are not able to render the water and escaping steam hotter than the single flame was able to render them. Why not? How much steam at 100 C. is required to raise the temperature of 54 ounces of water from 0P C. to 100 C., the latent heat of steam being taken = 540? 9. One pound of boiling water at 100 C. is placed in contact with.one pound of mercury at 0 C.; heat is transmitted from the water to the mercury, till both are at the same temperature. Calculate this temperature in degrees of the Centigrade scale. (Specific heat of...

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